The Thermodynamics of Linear Fluids and Fluid Mixtures by Pekař & Samohýl

## Exercise 1 to section $3.2^1$

The *rigid motion* can be expressed generally in the form  $\mathbf{x} = \mathbf{\Gamma}(t)\mathbf{X} + \boldsymbol{\gamma}(t)$ where  $\mathbf{\Gamma}(t)$  is the orthogonal tensor function of time (cf. exercise 3 to sect. 3.1). Find such change of frame functions  $\mathbf{Q}(t)$  and  $\mathbf{c}(t)$  (cf. p. 74) that the change of frame gives zero velocity  $\mathbf{v}$ . Try to answer before continuing reading.

The velocity transformation in the case of rigid motion is obtained substituting its above definition into (3.35):

$$\mathbf{v}^* = \mathbf{Q}\mathbf{v} + \dot{\mathbf{c}} + \dot{\mathbf{Q}}\mathbf{x} = \mathbf{Q}\mathbf{v} + \dot{\mathbf{c}} + \dot{\mathbf{Q}}(\mathbf{\Gamma}\mathbf{X} + \boldsymbol{\gamma}).$$
(1)

From definitions of velocity and motion, (3.7) and (3.1), respectively, it follows that  $\mathbf{v} = \underline{\dot{\chi}} = \mathbf{\dot{x}}$ ; upon combination with the general expression for the rigid motion we have:

$$\mathbf{v} = \dot{\mathbf{\Gamma}}\mathbf{X} + \mathbf{\Gamma}\dot{\mathbf{X}} + \dot{\boldsymbol{\gamma}}.$$
 (2)

Substituting (2) into (1) together with the requirement of zero transformed velocity results in:

$$0 = \mathbf{Q}\dot{\Gamma}\mathbf{X} + \mathbf{Q}\Gamma\dot{\mathbf{X}} + \mathbf{Q}\dot{\gamma} + \dot{\mathbf{c}} + \dot{\mathbf{Q}}\Gamma\mathbf{X} + \dot{\mathbf{Q}}\gamma$$
$$= (\mathbf{Q}\dot{\Gamma} + \dot{\mathbf{Q}}\Gamma)\mathbf{X} + \mathbf{Q}\Gamma\dot{\mathbf{X}} + \overline{\mathbf{Q}}\gamma + \dot{\mathbf{c}}$$
$$= \overline{\mathbf{Q}}\overline{\mathbf{\Gamma}}\mathbf{X} + \overline{\mathbf{Q}}\gamma + \dot{\mathbf{c}}$$
(3)

where we used the fact that time derivative of particle  $\mathbf{X}$  is zero.

Tensor **Q** is orthogonal, which, by definition, means that  $\mathbf{Q}\mathbf{Q}^T = \mathbf{1}$  and from this:

$$\overline{\mathbf{Q}\mathbf{Q}^T} = \mathbf{0}.$$
 (4)

Comparing (4) and (3) it is seen that (3) is fulfilled when  $\Gamma = \mathbf{Q}^T$  and  $\mathbf{c} = -\mathbf{Q}\boldsymbol{\gamma}$ . Thus, the answer to this exercise is as follows:

$$\mathbf{Q} = \mathbf{\Gamma}^T, \quad \mathbf{c} = -\mathbf{\Gamma}^T \boldsymbol{\gamma}.$$

<sup>&</sup>lt;sup>1</sup>Based on I. Samohýl: Irreversible Thermodynamics. Prague: University of Chemical Technology, 1998 (*in Czech*).