## Exercise 2 to section $3.8^{1}$

Using the barometric formula derived in exercise 1 to section 3.8, calculate the air pressure at 8 km above the Earth surface considering isothermal air at the (mean) temperature of $0^{\circ} \mathrm{C}$ and the mean molar mass of $29 \mathrm{~g} / \mathrm{mol}$. The pressure at the surface: 100 kPa . Compare the result with available (atmospheric, meteorological) data.

Answer. The integrated formula:

$$
P(h)=P(0) \exp (-M g h / R T) .
$$

Result: 36.71 kPa .

[^0]
[^0]:    ${ }^{1}$ Based on I. Samohýl: Irreversible Thermodynamics. Prague: University of Chemical Technology, 1998 (in Czech).

